\$i/019/60/000/010/007/057 h151/A029

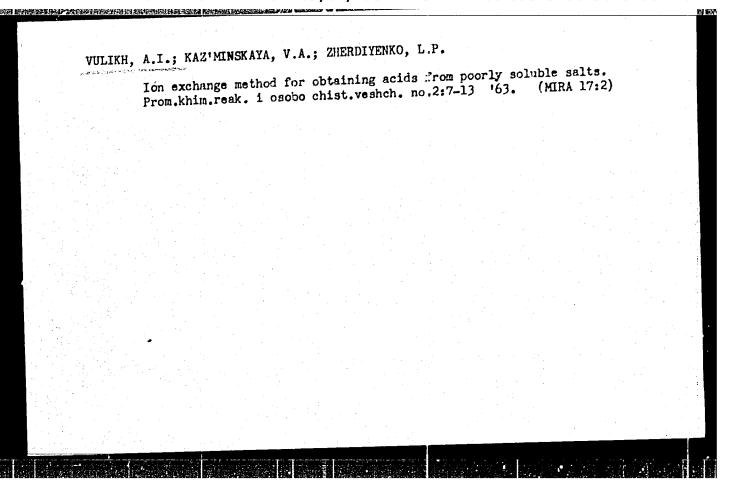
AUTHORS: Vulikh, A.I., Senyavin, M.M., Karoli, Yu.B., Korotkevich, B.I., Sidorova, L.G., and Galkina, N.K.

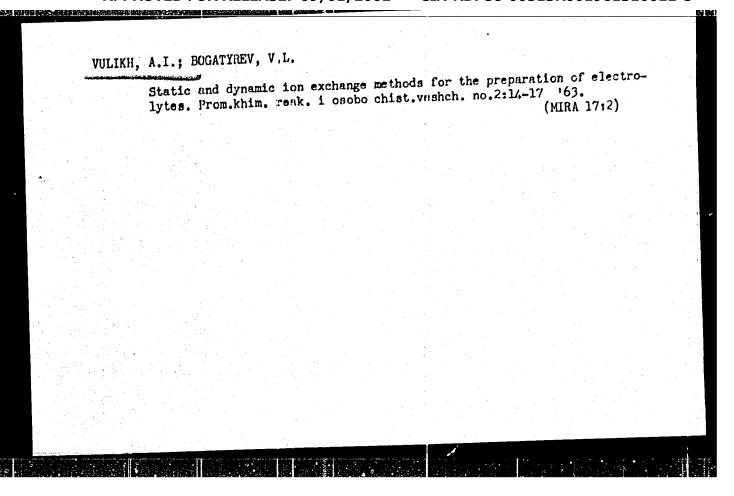
TITLE: A Method for Obtaining Rubidium or Cesium Carbonates

PERIODICAL: Byulleten' izobreteniy, 1960, Nr 10, p 19 Class 12 1, 16. Nr 128454 (622176/23 of Apr 21, 59).

TEXT: A method for obtaining rubidium or cesium carbonates from their dissolved salts, by a process of double ion-exchange. It has the following special features: to increase the degree of metal extraction, ammonium carbonate is used as the second reagent, whereupon the solution of rubidium (or cesium) carbonate and of ammonium carbonate is separated by distillation, while the solution or a salt separated therefrom is being heated.

Card 1/1



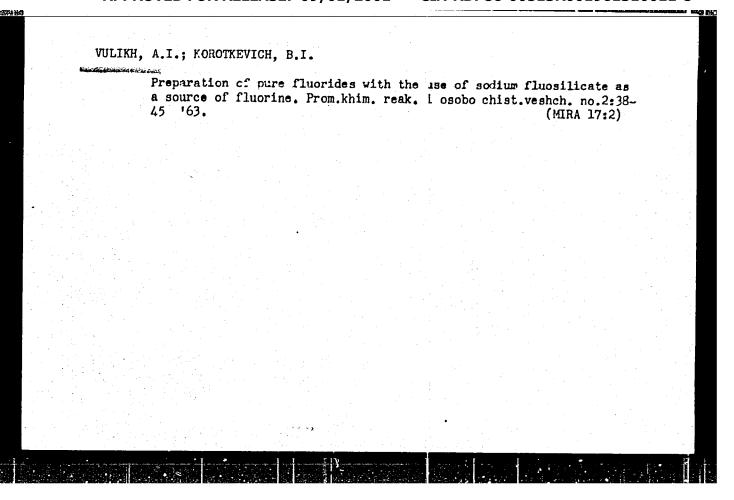


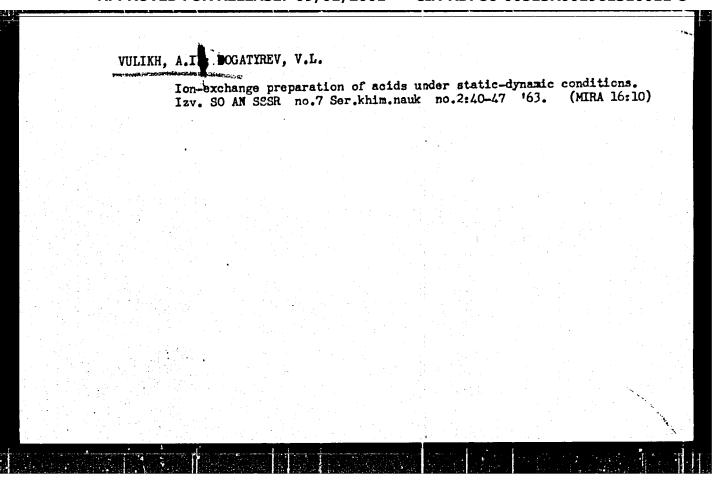
VULIKH, A.I.; STATSENKO, A.A.; MAKOVETSKIY, M.I.; MIL'SKIY, S.A.

Chemical method for the preparation of welding fluxes. Prom.khim. reak.
i osobo chist.veshch. no.2:18-22 '63. (MIRA 17:2)

ENG()/ENT(m)/SPIt(c)/EPF(n)-2/EPR/ENP(b) Pr-4/Ps-4/Pa-ASD(d)/AS(mp)-2/AND(m)-3/AEDO(b) JD/JG/MLK Accession Kr: AT4046116 8/0000/63/000/002/0023/0026 AUTHOR: Vulikh, A. I.; L.D. Prikhod ko; M. I. Kakiwatskiy TITIE: Preparation of subydrous lithium hydroxide and oxide from lithium hydroxide monohydrate SOURSE: USSR. Gosudarstvenny\*y komitet khimicheskoy i neftyanov promy\*shlennosti. Fromy\*shlennost' khimicheskiki reaktivov i osobo chisty\*kh veshchestv (Industry of chemical reagents and extra pure substances); informatsionnyty byulieten', no. 2. Moscow, IREA, 1963, 23 26 TOPIC TAGS: lithium hydroxide, lithium oxide, anhydrous lithium hydroxide, anhydrous lithium oxide, thermal decomposition, vacuum melting, vacuum dehydration, corundum crucible ABSTRACT: The thermal decomposition of lithium carbonate and lithium hydroxide monohydrate in a vacuum was investigated on a large scale, and the conditions for obtaining anhydrous lithium hydroxide and lithium oxide from the monohydrate were astablished. Among all the crucible materials tested, corundum was found to be the best for this purpose. A horizontal vacium electric furnace with a steel retort and Silit heaters was used, with a VN-2 oil racuum pump. The process was Card 1/2

L 12660-65 ACCESSION NR: AT40 6116			
at 300-350C in vacus (600-65) which is a porous product wi hydrate. In the second stag obtained at a gradually fucr decreasing pressure (down to cake, which separates readil lower than the theoretical y side the crucible. Thus, by lower than 900C, when the va ide can be avoided. Chemica 98-99% Li <sub>2</sub> O, less than 0.1%	O mm Hg) and the anhydrich a structure similar e, the complete dissoctions in the casing temperature (up the man Hg). The resulting from the corundum cruitald. No traces of the removing most of the por pressure of LiOH is I analysis showed that AI, and less than IX Li	o that of the initial mono- ition of lithium hydroxide is o 900-1000C) and a gradually ig lithium oxide is a solid lible. The weight is only 1-2% product could be detected out- iter from Lios at a temperature still low, loss of lithium ox- he reaction product contained	
hydroxide contained 0.5% CO <sub>2</sub> ASSOCIATION: None			
SUBMITTED: 27Roy63 NO REF SOY: 003	ENCL: 00 OTHER: 012	SUB CODE: IC, GC	
Card 2/2			

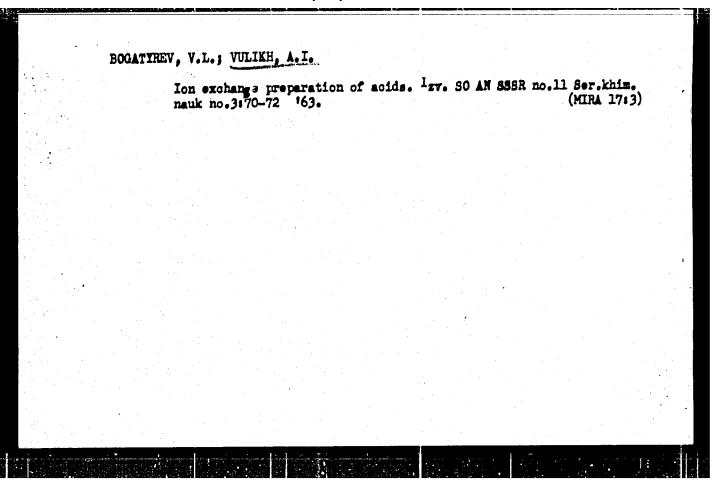




MIKULINSKIY, A.S.; KOZHEVNIKOV, G.N.; BAKHIREVA, L.D.; VULIKH, A.I.

Vacuum-thermal separation of cesium and potassium fluorides. Izv.
SO AN SSSR no.7 Ser.khim.nauk no.2;105-107 '63. (MIRA 16:10)

1. Ural'skiy filial AN SSSR, Sverdlovsk.



VULIKH, A.I., kand.tekhn.nauk; STATSENKO, A.A., inzh.; MAKOVETSKIY, M.I., inzh.; MIL'SKIY, S.A., inzh.

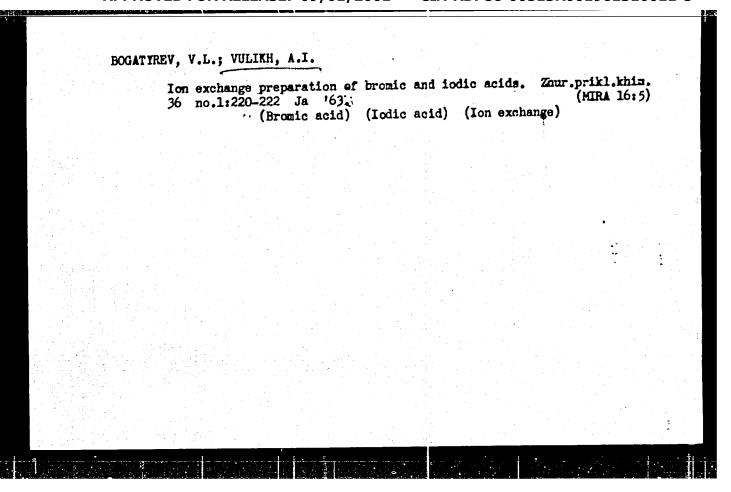
New technology for the production of fluxes for soldering and welding. Svar. proizv. no.9:24-26 S '63. (MIRA 16:10)

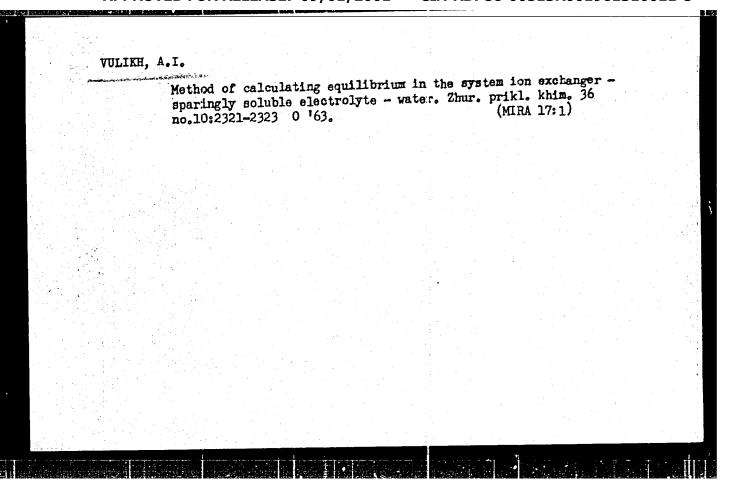
1. Novosibirskiy zavod khimicheskiki reaktivov.

Λ	CCESSION NR; AP4004065 S/0286/63/000/020/0012/0012
	UTHOR: Kutolin, S. A.; Vulikh, A. I.
N	o. 157967
	OURCE: Byul. izobret. i tovarn. znakov, no. 20, 1963, 12  OPIC TAGS: titanates, titanate synthesis, alkali metal titanates,
t	itanium oxides, titanium dioxide, arkara
	BSTRACT: An Author Certificate has been issued for a method of pre- aring alkali metal titanates by fusing alkali metal hydroxides with itanium dioxide at 1 mm Hg.
	ASSOCIATION: some SUBMITTED; 21Nov62 DATE ACQ: 13Dec63 ENCL: 00
	SUB CODE: CH NO REF SOV: 000 OTHER: 000

VULIKH, A.I. (Novosibirsk); KAZ'MINSKAYA, V.A. (Novosibirsk); ZHERDIYENKO, L.P. (Novosibirsk)

Chemical experiments with the use of ion exchangers. Khim. v shkole 18 no.5:57-65 S-0 '63. (MIRA 17:1)





NIKOLAYEV, A.V.; BOGATYREV, V.L.; VULIKH, A.I.

Study of ion exchange processes by means of physicochemical analysis.
Dokl. AN SSSR 153 no.2:360-362 N '63. (MIRA 16:12)

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR.
2. Chlen-korrospondent AN SSSR (for Nikolayev).

ACCESSION NR: AP4046449	/0076/64/008/010/2359/2361
AUTHOR: Kutolin, S. A.; Druz', N. A.; Vulikh	A. I.
: ITEE: Second stable modification of lithium m	etazirconate
SOURCE: Zhurnal neorganicheskiy khimii, v. 9	no. 10, 1964, 2359-2361
TOPIC TAGS: lithium metazirconate, stable m	dification, x ray analysis
ABSTRACT: The existence of a second stable nate in the Li <sub>2</sub> O-ZrO <sub>2</sub> system was established.  Limited furnace for 4 hours at 950C gave a production of that Li <sub>2</sub> Zr() <sub>3</sub> (I) synthesized by A. A. Griz peorg. khimii, 6, 2249 (1961)) from Li <sub>2</sub> CO <sub>3</sub> + 2 fours. Differences in the two modifications we	Heating of 2HOH 'ZrO2 in a ct LizZrO3 (II) which differd lk and V. Ye. Plyushchev (Zh. rO2 by heating at 1100C for 1-2 re estublished from x-ray data.
No mutual transitions of the two modifications we have call properties were established; fusion to 50C and Ll <sub>2</sub> ZrO <sub>3</sub> II, 1530 ± 50C; Censities4.1	nneratures-"LinZrOal, 1300 #
Grid 1/2	
Maria de la companya	

			TTEST I
ACCESSION NR: AP404	6449		
	materials was above 1.78.		
	d Li <sub>2</sub> ZrO <sub>3</sub> II hydrolyzed nac	h more readily than	
Li <sub>2</sub> ZrO <sub>3</sub> I. Orig. art. h	as: 2 figures and 1 table.		
ASSOCIATION: None			
SUBMITTED: 02Jul63	ENCL: 00		
SUB CODE: GC	NO REF SOV: 003	OTHER: 006	
Card2/2			
Barrager in complete the nation of the		The state of the s	

NIKOLAYEV, A.V.; BOGATYREV, V.L.; VULIKH, A.I. RT, CLT-H20. Zhur. neorg. khim. Ion-exchange system R<sup>+</sup>, IHI, 9 no.10:2469-2474 0 '64.

(MIRA 17:12)

1. Institut neorganicheskoy khimii Sihirskogo otdeleniya AN SSSR.

L 29:85-65 EM(m)/SWP(t)/SWP(b) WP(c) JD/JQ ACCESSION NR: AP5002192	B0/84/037/012/2748/2748
AUTHOR: Kutolin, S. A.: Vulikh A. I.  TITLE: Synthesis of alkali metal metatitanates und	//.
SOURCE: Zhurnal prikladnoy khimii, v. 37, no. 1	, 1964, 2748
TOPIC TAGS: alkali metal metatitanate, synthesis sium metatitanate	, lithium metatitanate, potes-
ABSTRACT: A method was worked out for the syn by sintering their hydroxides with TiO <sub>2</sub> under vacu	
was well mixed with a stoichiometric amount (Me <sub>2</sub> and placed in a vacuum furnace with a horizontal rebrought to 650 or 800C for the Li or K, respective	O:TiO <sub>2</sub> = 1) of LiOH or KOH
mospheric pressure. The pressure was then reduce temperatures were maintained for 2 more hours. product Li <sub>2</sub> TiO <sub>3</sub> was 1325 ± 50C and of K <sub>2</sub> TiO <sub>3</sub> , 82	ed to 0.5-1 inm Hg and the
Card 1/2	

"APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001961310011-8

L 234/5-65 ACCESSION NR: AP500219	2		
graphics			
ASSOCIATION: None			
SUBMITTED: 26Apr63	ENCL: 00	SUB CODE; GC	
NR REF SOV: 0//3	OTHER: 006		
Card2/2			

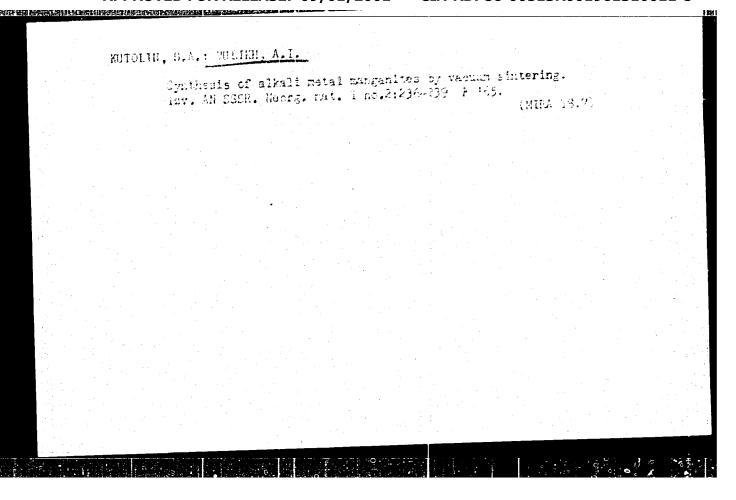
NIKOLAYEV, A.V.; BOGATYREV, V.L.: VULIKH, A.I.

Ion exchange system H°, Cell, || R', C1 H20 investigated by the ray method. Dokl. AN SSSR 155 no. 3:607.-610 Mr '64.

(MIRA 17:5)

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR. 2. Chlen-korrespondent AN SSSR (for Nikolayev).

L 35064-65 ENT(a)/EN	P(t)/EWP(E) IJP(c) JD/X	8/0286/65/	00/006/0019/0019
ACCESSION HR: AP500851	8 Arkhipov, S. M.; Sidorova.	引起 经银行证金	5
TITLE: A method for pr	oducing bromides and lodide	e of rubiaium	Imarità della colonia di California di California di California
	obreteniy i tovarnykh zrakov	no. 6, 1965,	19 compound, bromide,
lodide	agent, casium inorganic com		ducing bromides and
lodides of rubidium an	's Certificate introduces a d cesium by interaction of es of these metals with brouproduce only gaseous product he reducing agent improves	mine or iodine	tion. The use of
hydrazine hydiate as	하다 중앙소리는 강상은 한 100명 영향 100명 중앙국 본 중앙국 100일 등학 대기는 100년 대대 상태와 이 회장이라를 위한다고 있다는 만할 수		
ASSOCIATION: none SUBHITTED: 11May6 HO REF SOV: 000 Card 1/1	ENCL: 00 OTHER: CCO		SUB CODEL GC, IC



L 53696-05 ENT	(m)/EPF(m)-2/1/E-3/	EA(6)pu_4IJP(6),JD/6W/JG UR/0363/65/001/003/0388/0391
ACCESSION NR: APS		666.3:542.9
	Serceyeve	A. Ye. 57
AUTHOR: Kutolin,	S. A.; Vulikh, &. I.; Sergeyeve	at all supersals and the D
TITLE: Effect of	atmospheres of various gases or	the thermal synthesis and the
amountles of Maz	-Ka_03 and Well 08 thba combor	
COURCE: AN SSSR.	Izvestiya. Heorganicheskiye ma	tirialy, v. 1. no. 3, 1965, 388-391
	trate wised oxide salk	ali metal, titanium, zirconium,
it Pulliam alahimm.	AT ANT A LUIN LA MARSA MARANATA DE LA MARTINITA DE LA CONTRACTION DELA CONTRACTION DE LA CONTRACTION DEL CONTRACTION DE LA CONTRACTION DE	ad Michigan Anglichada (ili ad ici ad ili ali) (ili 🖺 un adilita e della completa Anglica)
	The second second	Tis He2 He TO3 and Ma He TO3 (where
ABSTRACT: Therma	NEW WINDS	and He is Nb , Ta ) was studied Preparation of these types of
in air and nitros	an atmospheres and single	leterated commic incustries
compounds is or I		LLE RESTEROCES VICE MATEL CASSOSIATES VICE
The title compour	nds were propared by tusing mixt ansition elements in the tempers	ture range from 50000 to 90000.  I tative formations of kn Ke 103 and
	varied from 1 to 8 hours. Quant	

$\sim Nb_2O_5$ , and $Ta_2O_5$ 1 the absence of min	eved by fusing carbonates of Linth the stream of nitrogen and in eralizate. Fresence of Sir atm	K. and Cs with TiO2; KrO2, HrO2, a vacuum at 700° to encoc and in sphere has a deleterious effect
on rate of formati	on of he we 0; and we we 0; un eaction products. The optimum mum reaction temperature is fro a vacuum (1 mm Hg). Densities, of saveral (1) le compounds wer	using duration is from 2 to 4
ASSOCIATION: non	ENCL: OG	SUB CODE 1 10, GC
NO REF SOV: 009	OTHERE OUT	
Card 2/2		

L 34503-65 ENT (m)/ENP(t)/ENP(b) LJP(c) JD ACCESSION NR: AP5002802	0078/65/010/001/0140/0144
AUTHOR: Kutolin, S. A. Vulkh, A. I.	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
TITLE: Synthesis of alkali metal metatitanate	in vacuum 7 7 - 1 1965, 140-144
SOURCE: Zhurnal neorganicheskoy khimii, v. TOPIC TAGS: alkali metal metatitanate, syntl	density, fusion
temperature, hydrolysis ABSTRACT: The reaction of LiOH, KOH and	1 12CO3 with TiO2 at atmospheric
pressure and under 1 him the same and under 1 him the same and alkali metal metalitanates was investigated, alkali metal metalitanates was investigated, alkali metali metalitanates was investigated, alkali metali metali metalitanates was investigated, alkali metali metalitanates was investigated, alkali metali metali metali metalitanates was investigated, alkali metali metali metalitanates was investigated, alkali metali metali metalitanates was investigated, alkali metali metalitanates was investigated, alkali metali metali metalitanates was investigated, alkali metali metali metalitanates was investigated, alkali metali	1 12TiO3 and K2TiO3 were produced  1 respectively) reaction under  1 red products and corrosion of the  1 red products and corrosion of the
corundum and corcelain crucibles. There wa and TiO, in air: under vacuum the reaction was and TiO, in air: under vacuum the reaction was showed the atructure of the hydroxide. X-rays showed the atructure of the contract of the coruntary of the	HERET WAS THE THE ALM GAMO OR WILL LIVE
Card1/2	

L 34503-65			
ACCESSION NR: AP5002802			
ty and the fusion temperature compounds were stable to 8			
ASSOCIATION: None	Ly than Lighton. C	ed in water, the K <sub>2</sub> TiO3 decom- orig. art. has: 2 tables and 1 figu	re.
SUBMITTED: 24Jul63	ENCL: 00	SUB CODE: GC, IC	
NR REF SOV: 005	OTHER: 016		
			100
			が決定
Card 2/2			

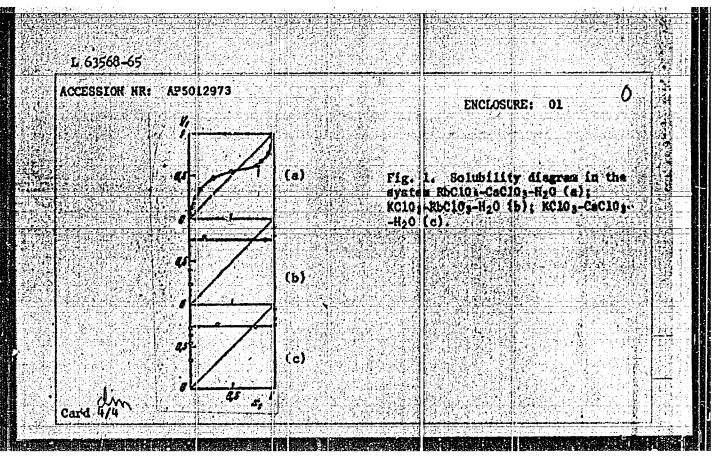
L 63568-65 EPA(&)-:/ENT(&)/EPF(:)/EPF(n)-2/T/ENF(t) Pt-7/Pu-1	/EMP(b)/B(4(b) Pr-4/ R/0078/65/010/005/1225/1228	
AUTHOR: Kirgintsev, A. N.; Kashina, N. I.; Vulikh, A	1.; Korotkevich, B. I. U	
TITLE: Ternary aqueous systems consisting of potassi	m, rubidium and cessum 27 17 17 17 17 18	NO. THE COURSE WATER CO.
ABSTRACT: The solubility of KC103-CsC103-H20, KC103-H2O ternary systems at 25°C is studied (see fig. 1 of tion of a feasible method for analyzing potassium, cerdiscussed. The method of isothermal desupersaturation bility. In the first system, no solid solutions were rate and cesium chlorate, and only the liquid phase was system, both the solid and liquid phases were analyzed formed between potassium chlorate and rubidium chlorate solid and liquid phase; were also stalyzed; rubidium colid and liquid phase; were also stalyzed; rubidium	the Enclosure). The selectum and rubidium systems is was used to study the solutormed between potassium chief analyzed. In the second in solid solutions were	
2ard 1/4		

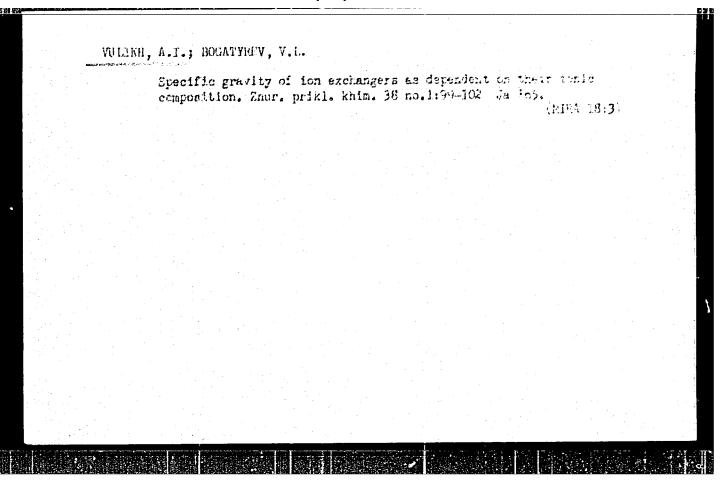
L 63568-65		
ACCESSION NRI AP5012973		
form a continuous series of solid solutions. The	e authors calculated the activ	
coefficients for RbClO3-CaClO3 solid solutions,	using F. Schmiling's formula	
lete = - feidig Dai: figit == -	Seedle Du.	
where $D_{\mathbf{x}} = \frac{z_{\mathbf{z}}  \mathbf{y}_{\mathbf{z}}}{y_{\mathbf{z}}  z_{\mathbf{z}}} \cdot D_{\mathbf{z}} = \frac{z_{\mathbf{z}}  \mathbf{y}_{\mathbf{z}}}{y_{\mathbf{z}}  z_{\mathbf{z}}}$		
The effect of composition on the logarithm of the ly described by equations from the theory of reg	e activity coefficient is adequate solutions	uate V
ln year 1,79 x2; ln ye=	170 zŁ	
The constant coefficient in these equations (1.7 with theory, this means that RbC103-CsC103 solid	solutions are close to phase	sepa-
ration. Orig. art. hau: 2 figures; 7 tables; a	nd 7 förmulas.	
Card 2/4		

"APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001961310011-8

ACCESSION NR: APJ012973		
ASSOCIATION: Institut neorg	miicheskoy khimii Sibii skogo	otdeleniya Akademii nauk
SSSR (Institute of Inorganic   SSSR)	Chemistry, Siberian Dipartm	ent. Acidemy of Sciences;
SUBMITTED: 04Dec61	ENCL: 01	SUB CODE: IC
NO REF SOV: 001	OTHER: 001	
Card 3/4		

"APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001961310011-8





VULIKH, A.I.; NIKOLAYEV, A.V.; ZAGOPSKAYA, M.K.; EOGATYREV, V.L.

Absorption of ammonia and chlorine by ion-exchange resins under dynamic conditions. Dokl. AN SSSR 160 nc.5;1072-1074, 7 '65.

(MIRA 18:2)

1. Institut neorganicheskoy khimii Sihirskogo otdeleniya AN SSSR.

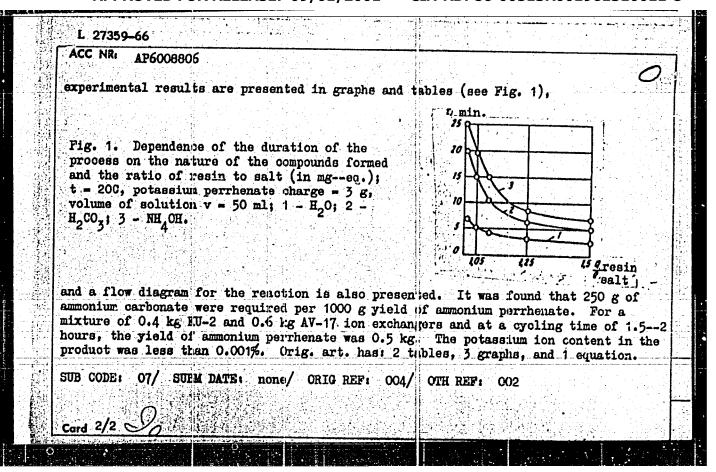
2. Chlen-korrespondent AN SSSR (for Nikolayev).

KIRGINTSEV, A.N.; AVVAKUMOV, Ye.C.; VULIKII, A.I.

Behavior of alkali metal impurities in the crystallization of cesium salts from a melt. Dokl. AN SSER 164 no.6:1315-1318 0 165. (MIRA 18:10)

1. Institut neorganicheskoy khimli Sibirskogo otdeleniya AN SSSR. Submitted March 29, 1965.

ENT(m)/ETC(f)/EWG(m) ACC NRI AP6008806 SOURCE CODE: UR/0136/65/000/011/0096/0099 AUTHORS: Bogatyrev, V. L.; Vulikh, A. I.; Sokolova, S. I. ORG: none TITLE: Derivation of ammonium perrhenate from potassium perrhenate with the aid of mixed bed ion exchangers SOURCE: Tsvetnyye metally, no. 11, 1965, 96-99 TOPIC TAGS: ammonium salt, rhenium compound, ion exchange : esin, cation exchanger, anion exchanger, ion exchange/ KU-2 cation exchanger, AV-17 anion exchanger ABSTRACT: This investigation was conducted to extend the work of N. M. Rubinshteyn (Avt. svid. No. 148390 (Byull. izobret., No. 13, 1962)). Ammonium perrhenate was synthesized from potassium perchenate and ammonium carbonate with the aid of a mixed bed KU-2 cation exchanger and AV-17 anion exchanger. The reaction was carried out according to the scheme  $RH + R'OH + KReO_4 = RK + R'ReO_4 + H_3O_7$  $2RH + R_2'CO_2 + 2KReO_4 = 2RK + 2R'R_1O_4 + H_2CO_2(H_2O + CO_2);$  $RNH_4 + R'OH + KReO_4 = RK + R'ReO_4 + NH_4OH (H_5O + NH_8t);$  $2RNH_4 + R'CO_8 + 2KReO_4 = 2RK + 2R'ReO_4 + (NH_4)_2CO_8(H_2O + CO_8) + NH_2).$ The optimum conditions for maximum yield of ammonium perrhenate were established. UDC: 669.849:66.074.7



#### "APPROVED FOR RELEASE: 09/01/2001

#### CIA-RDP86-00513R001961310011-8

L 34610-66 EWT(1)SOURCE CODE: UR/0210/66/000/003/0100/0102 ACC NR: AP6026571 AUTHOR: Vulikh, A. I. (Candidate of technical sciences); Shivandronov, Yu. A. (Candidate of technical sciences); Zagorskaya, M. K. (Candidate of technical sciences); Bogatyrev, V. L. (Candidate of chemical sciences) ORG: Novosibirskiy Factory of Chemical Agents (Novosibirskiy zavod khimicheskikh reaktivov); Institute of Inorganic Chemistry, Siberian Branch, AN SSSR (Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR TITIE: Filtering ionite gas mask SOURCE: Gigiyena i sanitariya, no. 3, 1966, 100-102 TOPIC TAGS: gas mask, gas absorption, ion exchange resin, gas mask component, gas filter, industrial hygiene ABSTRACT: The authors tested in a wide range of concentrations and gas velocities the absorption from gas-air mixtures of armonia, amines (by KU-2 cationite in Tylengen form), sulfur dioxide, chiorine, and hydrogen chloride (by AV-17 and EDE-10P anionites in the hydroxyl and carbonate forms). The basic and acidic gases were invariably completely absorbed. The capacity of the ionites was 80-90% of the total exchange capacity, 1.e., 4 meq/g for KU-2 and about 3 meq/g for AB-17. The mist universal absorbents are the highly ionized single-function resins (KU-2, OBS-3, SEV, and AV. The carboxyl cationites (e.g., KB-4) and anionites with secondary and tertiary UDC: 614 **Card** 1/2

#### L 34610-66

ACC NR: AP6026571

amino groups (e.g., EDE-10P), whose capacity is 8-9 geq/kg, seem to be more effective in absorbing strongly acidic and strongly basic gases. Ionites with large pores (KU-2P for amines, etc.) are best for absorbing gases or fumes of organic substances with large molecules.

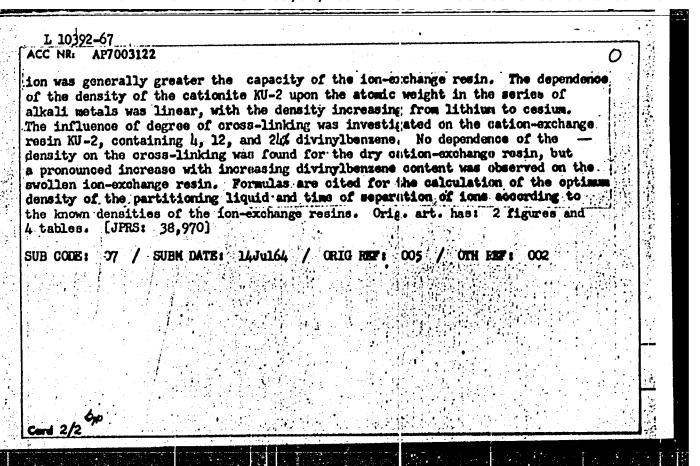
The article concludes with a brief description of an ionite gas mask successfully used for several menths under industrial conditions to provide protection against associate. An antidust filter from a RP-5 respirator is mounted on the lower part of the tank. KU-2 in the II form was the absorbent used. The total weight of the tank with the antidus: filter was 200-250 g. loaded with 50 g of KU-2, it absorbed 3.5 g of associate and worked continuously for 30 hours. Orig. art. has: 1 figure and 1 table. [JPRS: 36,455]

SUB CODE: 06, 15, 07 / SUBM DATE: 24Nov64 / ORIG REF: 003 / OTH REF: 001

Card 2/2/0

JD/JG/WB L 05025-67 EWT(m)/EWP(t)/ETI IJP(c) ACC NR: AP6032980 SOURCE CODE: UR/0078/66/011/010/2328/2330 AUTHOR: Kirgintsev, A. N.; Avvakumov, Ye. G.; Vulikh, A. I. ORG: Institute of Inorganic Chemistry, Siberian Branch, AN SSSR (Institut neorganicheskoy khimii, Sibirskoye otdeleniye, AN SSSR) TITLE: Cesium nitrate purification by zonal recrystallization , SOURCE: Zhurnal neorganicheskoy khimii, v. 11, no. 10, 1966, 2328-2330 TOPIC TAGS: metal crystallization, recrystallization, oriented crystallization, alkali metal, cesium nitrate, zonal recrystallization ABSTRACT: The method of oriented crystallization is used to determine the distribution of alkali metals in cesium nitrate at different crystallization rates (under constant mixing). The data obtained show that the method of zonal recrystallization may be recommended to free cesium nitrate of alkali metals. Orig. art. has: 1 table and 3 figures. [Authors' abstract] SUBM DATE: 08Jan65/ ORIG REF: 005/ SUB CODE: 07/ Card 1/1 UDC: 546, 36'175:548, 53

L 10392-67 ENT(IL) DS/RM ACC: NR. AP7003122	SOURCE CODE:	UR/0080/6:, 139/00	6/1760/1765
AUTHOR: Bogatyrev. V. L.; Vulikh, ORG: Institute of Inorganic Chemis	A. I.; Sokolova, S. try, SO, AN SSSR (I	I. natitut neorganiche	skoy khimii
SO AN SSSR) TITIE: Density of <u>ion-exchange res</u>	ins 1		
SOURCE: Zhurnal prikladnoy khimii, TOPIC TAGS: ion exchange resin, po			
ABSTRACT: A systematic determination widespread industrial cation—and a (in the dry and swollen states), for analytical methods based on the use the density of the investigated ior ion, grain size, and degree of cross the temperature within the range is upon the results of the determination observed: 1) the density of the state than the density of the dry resin, than one in all cases; 2) the density corresponds to the ratio of the determination of the density of the same ion—exchange resin increases sorbed ion; 4) the density of variances.	or use in the development of ion-exchange resint upon the control of the cation of the	opment of technologies in s. The dependence on the nature of the instracted. Fluctuate have no significant erns of variation we resin was always lead the latter was gently and exchange resins was anion-exchange resitrices; 3) the dence of the latter was gently alent weight	ical and nce of he sorbed ion of effect ere ess reater generally ins, which sity of
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ACC NR: AP6032947

SOURCE CODE: UR/0363/66/002/010/1803/1810

AUTHOR: Kutolin, S. A.; Vulikh, A. I.; Druz', N. A.; Shammasova, A. Ye.

ORG: none

TITLE: Dependence of the structure and properties of the A2BO3 and ABO3 compounds on the composition of the gaseous atmosphere in thermal

synthesis

SOURCE: AN SSSR. Izvestiya. Neorganicheskiye materialy, v. 2, no. 10, 1966,

1803-1810

TOPIC TAGS: ferroelectric material, antiferroelectric material, dielectric

constant, physical chemistry property, refractive index

ABSTRACT:

in a recently published article, the authors [association unknown] analyzed the data from Western and Soviet literature, including their own experimental data which were published in 1964-66, on the thermal synthesis, structure, and properties of A2BO3 and ABO3 compounds, where A is an alkali metal and B is Ti, Zr, Mn, Nb, or Ta.

In previous publications, the authors established the effect of the gaseous medium in which the compounds were synthesized on their structure and particle size. Now, they have made a detailed analysis of the earlier data to correlate the conditions of synthesis,

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primarily the gaseous medium, with the physicochemical properties of the compounds. In the authors' opinion, this analysis is of practical importance for the synthesis and application of these compounds. The properties studied were: density, index of refraction, dielectric constant, intensity of IR absorption bands, and catalytic activity. The experimental data were obtained with samples sintered at a relatively low temperature from a solid mixture of an alkali carbonate and an acidic oxide, in vacuum or in a nitrogen stream.

The nature of the gaseous medium was shown to affect only the structure of alkali metatitanates and manganites (A<sub>2</sub>BO<sub>3</sub>), and not their physicochemical properties, such as density, index of refraction, or dielectric constant. Density was the only property of the manganites which was actually measured; the index of refraction and dielectric constant of the manganites exceeded the measurable values. An exception was the crystal symmetry of K<sub>2</sub>TiO<sub>3</sub> and RbTiO<sub>3</sub> which apparently remained unaffected by the gaseous medium in which their synthesis was accomplished. However, the existence in these two compounds of second order phase trunsitions, undetected by x-rays, may not be excluded. In all alkali metatitanates the intensity of the IR absorption bands due to deformation vibrations of

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ACC NR. AP6032947

the [TiO<sub>6</sub>] octahedra was found to be independent of the method of synthesis. Their catalytic activity was affected by the gaseous medium, as shown, for example, by the comparative data on specific surface, preexponential factor, and activation energy for a maximum decomposition of hydrogen peroxide on a Li<sub>2</sub>TiO<sub>3</sub> catalyst prepared in the air or in vacuum.

In the group of A<sub>2</sub>BO<sub>3</sub> and ABO<sub>3</sub> compounds, where B is Zr, Nb, or Ta, i.e., alkali metazirconates, metaniobates, and metatantalates, only NaTaO<sub>3</sub> behaved like the alkali metatinates and manganites versus the gaseous atmosphere in the synthesis. The gaseous atmosphere changes the crystal structure, i.e., symmetry type and lattice constants of NaTaO<sub>3</sub>, but does not affect its picnometric density or intensity of deformation vibrational bands in their IR transmission spectra. Other compounds of this group -- Li<sub>2</sub>ZrO<sub>3</sub>, NaNbO<sub>3</sub>, KNbO<sub>3</sub>, CsNbO<sub>3</sub>, and CsTaO<sub>3</sub> -- change their crystal structure, i.e., symmetry type and/or lattice constant, in different gaseous media simultaneously with certain physicochemical properties, e.g., picnometric density, dielectric constant, intensity of deformation vibrational bands in the IR absorption spectra, and catalytic activity versus H<sub>2</sub>O<sub>2</sub> decomposition.

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## ACC NR. AP6032947

The crystal structure of LiNbO<sub>3</sub>, LiTaO<sub>3</sub>, and KTaO<sub>3</sub>, was not affected by the difference in gaseous atmosphere in the synthesis, but picnometric density, index of refraction, and intensity of deformation vibrational bands of the IR spectra were substantially changed.

These diverse and strong effects of the gaueous medium on the structure and properties of  $A_2BO_3$  and  $ABO_3$  compounds were explained as the result of deformability of their structure, specifically of the rendency toward distortion of the  $[TiO_6]$ ,  $[NbO_6]$ , and  $[TaO_6]$  octahedra. This deformability was correlated with a significant ionic polarizability of the alkali metatitanates, raetaniobates, and metatantalates. This correlation which was experimentally established for the above-indicated compounds (presumably) may be extended to other compounds with significant ionic polarizability and may form the base for predicting the possibility of a beneficial effect of a given gaseous medium on the completeness of synthesis of a given compound. In addition, a significant ionic polarizability of a given compound may be an indication of a potential ferroelectric or antiferroelectric property.

An additional indication of the possible ferroelectric or antiferroelectric property of alkali metatitanates was seen in the ob-

ACC NR. A16032947

served analogy in the structure of their IR absorption bands which are linked to the stretching vibrations of the [TiO<sub>6</sub>] octahedra and in the structure of the corresponding bands of the [NbO<sub>6</sub>] and [TaO<sub>6</sub>] octahedra in the IR abosrption spectra of the alkali metaniobates and metatantalates. The observed spectral structure is characteristic of ferroelectric materials. The authors concluded that confirmation of the effect of a gaseous medium on solid-phase synthesis of a given compound is a prerequisite for studying the ferroelectric property in this compound. Orig. art. has: 1 figure and 8 tables.

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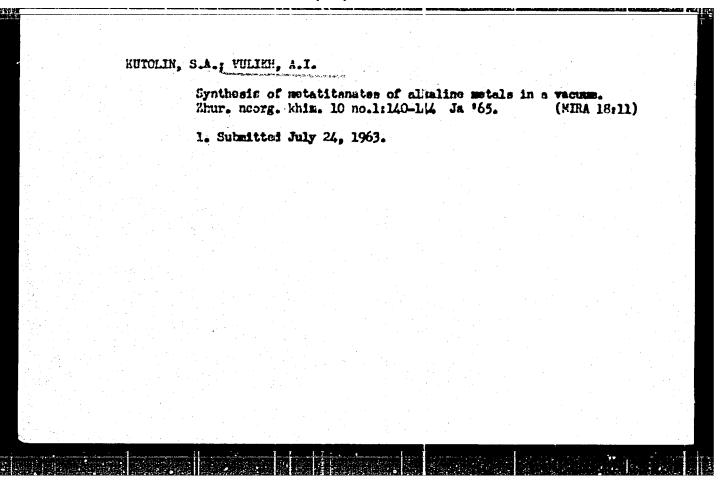
Card 5/5

BOUATYREV, V.L.; VULIKH, A.I.; SOKOLOVA, S.I.

Obtaining ammonium perrhenate from potassium perrhenate with the help of a mixed layer of ionites. TSvet. met. 38 no.11: 96-99 N \*65.

(MIRA 18:11)

	L 10846-66 EWT(m)/ETC/EWS(m) # RW/DS
	ACC NR: AP6000233 SOURCE CODE: UR/0289/65/000/002/0023/0027
	AUTHOR: Nikolayev, A. V.; Bogatyrev, V. L.; Vulikh, A. I. 1913  ORG: Institute of Inorganic Chemistry, Siberian Section, AN SSSK, Novosibirak B (Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR)
	TITLE: Separation of cation and anion exchangers in organic liquids
	SOURCE: AN SSSR. Sibirskoye otdeleniye. Izvestiya. Seriya khimicheskikh nauk, no. 2, 1965, 23-27
	TOPIC TAGS: anionite, ion exchange resin
	ABSTRACT: The cation exchanger KU-2 was separated from the anion exchanger AV-17 in mixtures of benzene, dichloroethane, and carbon tetrachloride. Values of the density and viscosity at 20C in these systems were determined. The dependence of the time of separation was studied as a function of the density of the separating liquid and grain size of the exchangers, and the effect of the difference in the density of the cation and anion exchanger during their separation was demonstrated. Formulas derived earlier for the calculation of the optimum density of the separating liquid and duration of separation of cation and anion exchangers were confirmed experiment-
	ally. Orig. art. has: 3 figures and 5 tables.
	SUB CODE: 07, 11 / SUBM DATE: 15Jun64 HU Card 1/1 UDC: 541.13
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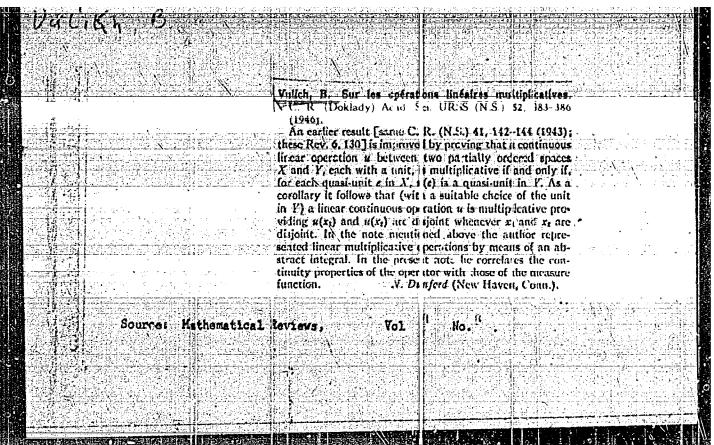
## VULTKH B.L. ) ...

Nekotoryye teoremy o posledovatel'nostyakh razryvnylh funktsiy. DAN, 1 (1935), 357-362. Ob odnom tipe metricneskikh prostranstv. DAN, 4 (1915,) 295-298. Sur les forces ge'ne'rals de certaines ope'rations line'aires. Matem. sb., 2 (44). (1937) 275-306. K hopmirovannyye prostranstva. L., Uche'in. zap. ped. in-ta, 28 (1939), 179-224.

So: Mathematics in the USSR, 1917-1947 adited by Kurosh, A.G., Narkushevich, A.I., Rashevshiy, P.K. Moscow-Leningrad, 1948

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The function $u$ is partial whenever $x_t$ and $x_t$ and $u(x_t)$ are disjoint	ly additive if $u(x_i + x_i) = u(x_i) + u(x_i)$ for disjoint; it is disjoint if $u(x_i)$ whenever $x_i$ and $x_i$ are. The func-	(ADC) u(x) = on eac	respectively) is given by the Radon interr $Y(\lambda)d\varphi(\alpha)x$ , where $Y(\lambda)$ is uniformly continuous
The function $u$ is partial whenever $x_t$ and $x_t$ and $u(x_t)$ are disjoint tion $u$ is uniformly of $\lim_{t\to 0} \{u(x_t) - u(x_t)\}$ if	by additive if $u(x_i + x_i) = u(x_i) + u(x_i)$ ire disjoint; it is disjoint if $u(x_i)$ whenever $x_i$ and $x_i$ are. The func- itinuous if, for every constant $C_i$ zero in the $(o)$ -topology where	(ADC)  u(x) =  on each function  If in ac	respectively) is given by the Radon interversely $(AD)^2 = Y(\lambda) d\varphi(a)x$ , where $Y(\lambda)$ is uniformly continuous finite segment whose values lie in $Y$ , $\varphi$ is a unitary (a measure function, respectively) and $Y(0) = (AD)^2 = (AD)$
The function $u$ is partial whenever $x_i$ and $x_i$ and $u(x_i)$ are disjoint tion $u$ is uniformly of $\lim_{t\to 0}  u(x_i) - u(x_i) $ if $ x_i - x_i  < \delta \cdot 1$ and $ x_i $ disjoint uniformly continued.	ly additive if $u(x_i + x_i) = u(x_i) + u(x_i)$ are disjoint; it is disjoint if $u(x_i)$ whenever $x_i$ and $x_i$ are. The func- itinuous if, for every constant $C$ ,  zero in the $(o)$ -topology where $ x_i  < C \cdot 1$ . A partially additive through operator is called an operator	(ADC) u(x) = on each function If in actinuous	respectively) is given by the Radon integral $X = Y(\lambda) d\varphi(a)x$ , where $Y(\lambda)$ is uniformly continuous finite segment whose values lie in $Y$ , $\varphi$ is a unitar $Y(\lambda) = (x + y)$ , where $Y(\lambda)$ is uniformly continuous to measure funition, respectively) and $Y(0) = (x + y)$ is uniformly contact finite segment.
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The function $u$ is partial whenever $x_t$ and $x_t$ and $u(x_t)$ are disjoint tion $u$ is uniformly collimated $ u(x_t)-u(x_t) $ if $ x_t-x_t  < \delta \cdot 1$ and $ x_t $ disjoint uniformly contained $(ADC)$ . If in addition the	ly additive if $u(x_i + x_i) = u(x_i) + u(x_i)$ are disjoint; it is disjoint if $u(x_i)$ whenever $x_i$ and $x_i$ are. The func- itinuous if, for every constant $C$ ,  zero in the $(o)$ -topology where $ x_i  < C \cdot 1$ . A partially additive through operator is called an operator	(ADC) u(x) = on each function If in act tinuous Friche	respectively) is given by the Radon integral $X = Y(\lambda) d\varphi(a)x$ , where $Y(\lambda)$ is uniformly continuous finite segment whose values lie in $Y$ , $\varphi$ is a unitar $Y(\lambda) = (x + y)$ , where $Y(\lambda) = (x + y)$ and $Y(\lambda) = (x + y)$ is uniformly continuous to measure function, respectively) and $Y(\lambda) = (x + y)$ is uniformly continuous each finite segment, then for each $x$ in $X$ the differential $d_x(x)$ of $u$ exists. $N$ . Dunford,

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Vulih, B. Z. The product is lister partially andered spaces and its application to the fixory of operations. L. Mat. Sbornik N.S. 22(64), 27-18 (1948). (Russian) Yulih, R. Z. The product in linear partially ordered spaces and its application to the theory of operations. IL. Mat. Shornile N.S 22(64), 267-317 (1948). (Russian) The author presents a detailed exposition, with a number of extensions, of results previously arnounced IC. R. (Doklady) Acad. Sal. WRSS (N.S.) 16, 850-854, 855-859 (1940); 41, 142-144, 187-190 (1943), 52, 95-98, 383-386, 475-473 (1946); these Rev. 2, 221, 222; 6, 130, 8, 468; 9, 41]. Close connections exist between the theory expounded here and earlier results obtained by Freudenthal, Kantorovich, M. and S. Krein, and Kakutani [Freudenthal, Akad Wetensch, Amsterdaus, Proc. 39, 641-651 (1936); Kantorovitch, Rec. Math. [Mat. Sbornik] N.S. 2(44), 121-168 (1937); 7(49), 209-28£ (1940); M. and S. Krein, Rec. Math. [Mat. Sbornik] N.S. 13(55), 1-38 (1943); Kakutani, Ann. of Math. (2) 42, 523-537, 994-1024 (1941); these Rev. 2, 317; 6, 276; 2, 318; 1, 205]. Let E be a partially ordered linear space over the real numbers (which are denoted by R in the sequel) of type  $S_1$ : i.e., (1) for some  $x \neq 0$ , O<x; (II) O<x and O<y imply O<x+x; (III) for every reX, there exists an kiel such that x1-x>0; (IV) 0 <x;

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Note, a further restriction is placed upon the spaces considere i. A positive element in X, which may be denoted by the symbol 1, is said to be a unit if  $\inf(x, 1) > 0$  for all positive xeX. It is supposed that X contains a unit element and t ext this unit is fixe I once and for all. An element exY is said to be unitary if  $\inf(x, 1-x)=0$ , and the set of all unitary elements in X is fenoted by  $\mathfrak{C}(X)$ . It is proved that unitary elements behave generally like projection operators in Hilbert space. For every xeX, let  $e_x$  (the "characteristic" of x) be the least element in  $\mathfrak{C}(X)$  with the property that  $\inf(\{x\}, 1-e\}=0$ ; it is shown that  $e_x$  exists for all xxX and that  $e_x$  expandinf  $\{\pi\{x\}, 1\}$ . Various other formal properties of  $e_x$  it e also established.

Note; for all  $x \ge 0$ , let S(x) be the set of all states  $\sum_{i=1}^n a_i e_{i,i}$  where the  $a_i$  are managative real numbers,  $e_i \in S(X)$ , and the entire sum is less than of equal to x. For x, y in X and homegative, consider the set B of all sums  $\sum_{i=1}^n a_i e_i$  in  $(e_i, e_i)$ , where  $\sum_{i=1}^n a_i e_i x S(x)$  and  $\sum_{i=1}^n a_i e_i x S(y)$ . If B is bounded above the element suppose x is defined to be the product xy where  $\sum_{i=1}^n a_i e_i x S(x)$  and y. If B is unbounded, then the product xy where  $\sum_{i=1}^n a_i e_i x S(x)$  is unbounded, then the product xy is defined to be the product xy.

Source: Nathemetical Reviews,

Vol 10.

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does not exist. For a and I nonpositive, his product y, title fined as sup (x, 0), sup (j, 0) + (-1) int (z, 0), (-1) int (y, 0) - sup (x, 0), (-1) int (y, 0) - (-1) int (d, 0, asp (y, 0)) (sali

The next topic treated is that of spaces & which are ring under the operations it +y and xy. An element ket is said to be bounded if |x| = CI |op some GeR; the set of all such the set of all segments is denoted by |x|. Then ket clearly a space of type Sa; |x| |x| = x ing; and it admits the norm |x| = inf |E| = x |x| = x |x|.

||x| = inf |E| C, CeR, |x| | |x| = C |x|. Then |x| = x |x| = x |x| = x |x|.

Source: Mathematical Reviews

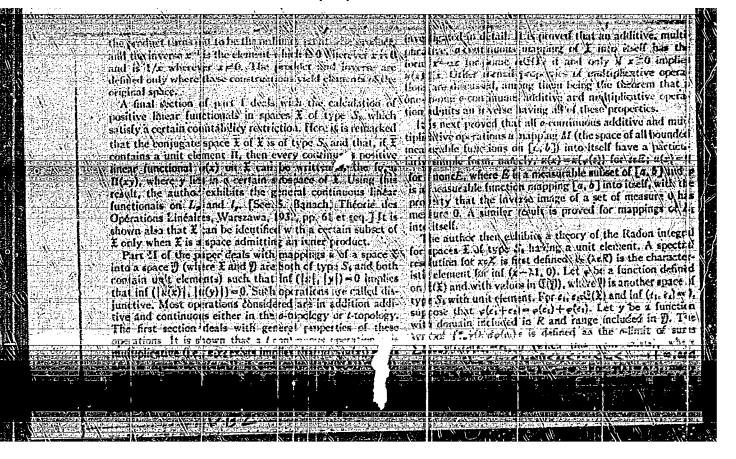
by a number of authors [see, for example, I. M. Gelland, lee, Ma I. [Mat. Shoralk, M.S. 9(51), 5-25 (1941); these Rev. 3 [1] It is identifiable with a ring of real-valued dontines as functions defined on a certain compact lines, don't space it. I. [Reviewer's note: X<sub>0</sub> is identifiable with the set of all feal-valued continuous functions defined on Y if and both it is is complete in its notern and if this is the said and of this is the said and the said and this is said and of this is the said and the said and the said and said and the said and said and the said and with the existing of inverse cless as follows. If for an xxX, there exists yell such that the exist and is denoted by a first inmediate that exist for all and is denoted by a first inmediate that exist for all beroff the usual formal properties of inverses in commentative rings are established, and it is shown that, the product is an occurrence.

The artific next considers the exist energy of inverse to the artific next considers the exist yell such that the exist yell and only if it is complete in its norm; and if this is the case, injury a very strong disconnectivity property, in like the property of property and only if it is complete in its norm; and if this is the case, injury a very strong disconnectivity property, in like the property in the case, injury a very strong disconnectivity property, in like the exist yell and only if it is complete in its norm; and if this is the case, injury a very strong disconnectivity property, in like the property of like only is complete in its norm; and if this is the case, injury a very strong disconnectivity property, in like the property of like of the property of like of the like the case, injury a very strong disconnectivity property, in like the like the case, injury a very strong disconnectivity property, in like the like the case, injury a very strong disconnectivity property, in like the case, injury a very strong disconnectivity property, in like the case, injury a very strong disconnectivity property, in like the case, injury a very strong disconnectivity property, in like the case, in like the case of the c apply, it course, to Gatich spaces suffering Kakutim's axioms for an Afspace Described Much of the interest of the prefent paper lies in the fact that multiplication is defined without recourse us a representation theory.]

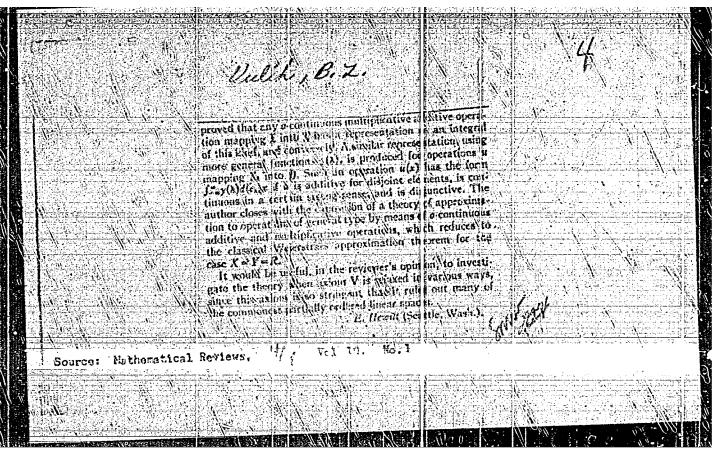
A in latter of applications of the constructions contined above to discussed to E. I. and the space of all country by infinite sequendes of real numbers. In each case,

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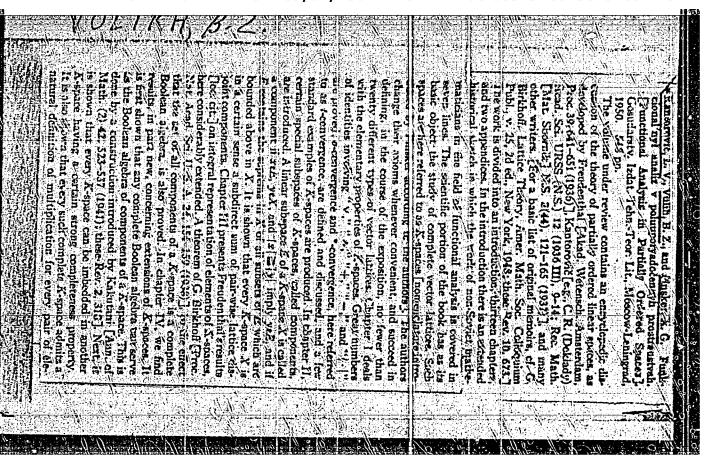
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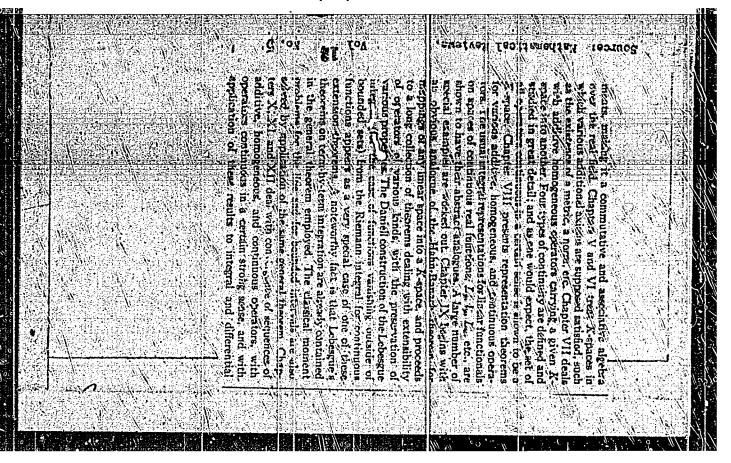


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Dokazatelstvo Odnoy Teoresy Kreynor. Uchen Zapiski (Demipr. Gos. Ped. In-T IM Gertsena) T. LXIV, 1948, S. 9-15

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equations. Here the principal tools are a number of fixed point theorems for abstract operators. Chapter XIII presents a survey of the known results concerning the concrete representation of K-spaces, as well as some new larts contemporator representation of K-spaces, as well as some new larts contemporate representation of K-spaces, as well as some new larts contemporate representation of K-spaces, as well as some new larts contemporate representation of K-spaces, as well as some new larts contemporate that the space in question contains with x all y such that  y  \( \frac{1}{2} \)  x . It is also shown that \( \frac{1}{2} \) spaces. The linear subspace in question contains with x all y such that  y  \( \frac{1}{2} \)  x . It is also shown that \( \frac{1}{2} \) spaces. It is book seems to such from a number of shortcoming. First, as to omissions \( \frac{1}{2} \) where a therefore the subspace in question contains with x all y such that \( \frac{1}{2} \) spaces. It is book seems to such from a number of shortcoming. First, as to omissions \( \frac{1}{2} \) where a there given a definition of the weak topology, about which are surfaced by some interesting facts could be reacheded that the property of the first and the spaces. The limit is the property of the spaces and the proliferation of detentions and animber of shortcoming. First, as to omissions \( \frac{1}{2} \) where as there are all the property seamed to be the great the legislations to analyses, which are surfaced to be the great the proliferation of detentions and animals is so great that concentrated the research what has been contained to be the great the proliferation of detentions and animals are surfaced by the most surface and the applications to analyses, which are surfaced to be the great the limit of the theory seems for the most surface and the proliferation of the work and the applications to analyses, which are surfaced to be the great the limit of the proliferation of the works and the proliferation of the proliferation of the works and	, , <u>, , , , , , , , , , , , , , , , , </u>	81 Tog	, and then Inc	t exted tail	reamos
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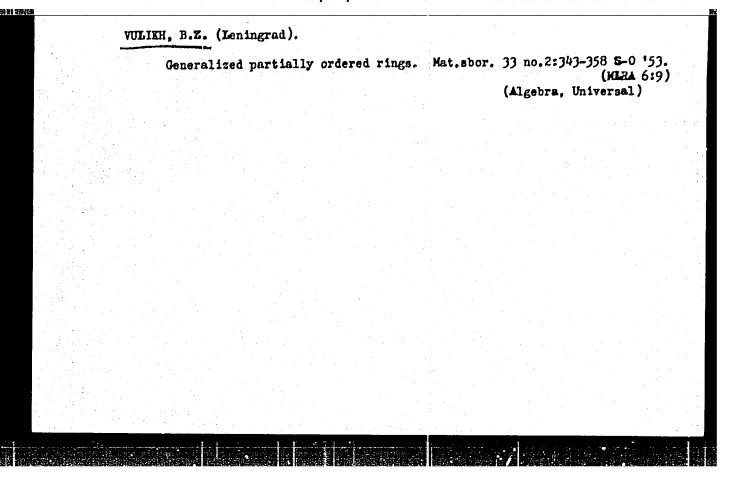
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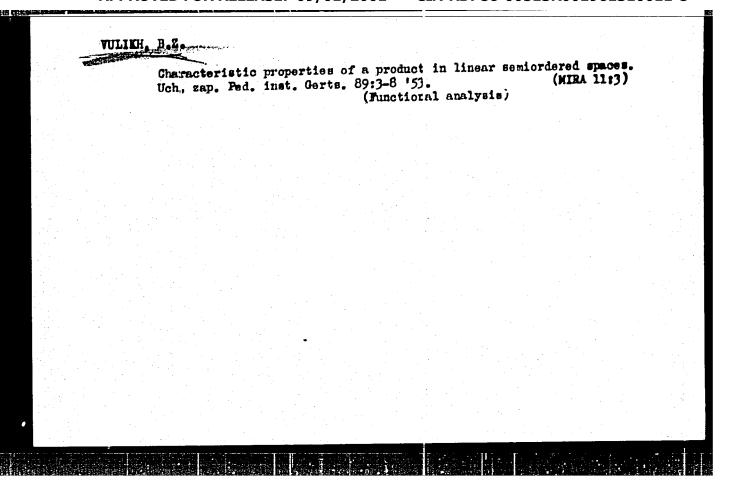
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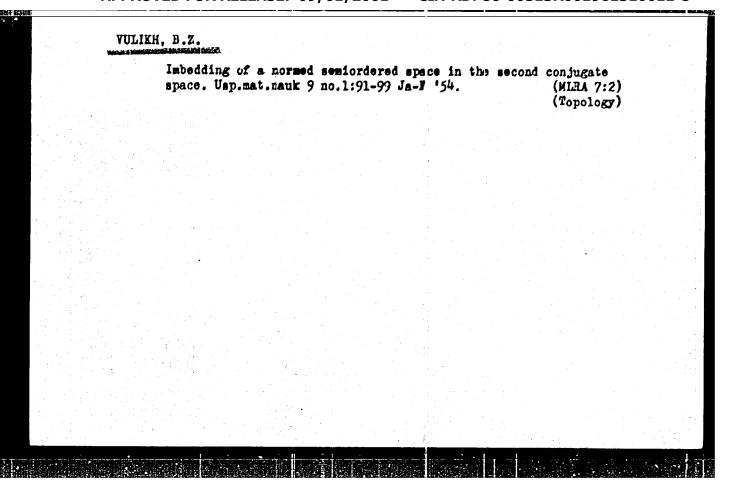
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System

USSR/MATHEMATICS/Functional analysis

CARD 1/1 PG - 750

AUTHOR

VULICH B.Z.

TITLE

The application of the theory of partially ordered spaces to the investigation of selfadjoint operators in the Hilbert

space.

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Uspechi mat. Nauk 12, 1, 169-172 (1957)

reviewed 5/1957

The great analogy of the proofs and results of the theory of partially ordered spaces (lattice theory) and the theory of selfadjoint operators induces the author to apply directly the theory of partially ordered spaces to the investigation of the selfadjoint operators. Without proof beside of some own results the author enumerates several well-known results of Stone, Sobolev, Ljubowin etc. Most of the results relate to bounded operators.

VULIEN, B.Z.

Partial order in rings of bounded self-conjugate operators [with summary in English]. Vest, LGU 12 no.13:13-21 '57. (MIRA 10:11) (Operators (Mathematics))

PHASE I BOOK EXPLOITATION

656

#### Vulikh, Boris Zakharovich

Vvedeniye v funktsional'nyy analiz (Introduction to Functional Analysis) Moscow, Gos. izd-vo fiziko-matematicheskoy lit-ry, 1950. 352 p. 7,500 copies printed.

Ed.: Akilov, G. P.; Tech. Ed.: Volchok, K. M.

PURPOSE: This book is intended for those interested in the fundamentals of functional analysis who do not have previous training in the more specialized branches of mathematics, and it may also be useful to engineers.

COVERAGE: The author gives only the fundamentals of functional analysis and their applications to problems in various fields of muthematics. Therefore the basic concepts of Euclidean, metric, normed, Hilbert and L. spaces are given. A short theory of operators in various spaces, as well us their properties, are presented. As particular applications of functional analysis, the basic theorems of integral equations are presented. Various problems are studied, such as the problem of the best approximation, boundary value problem of differential equations, approximate methods of the solution of equations, generalized methods of the summability Card 1/10

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No personalities are mentioned.	21 -0/200 1020101			
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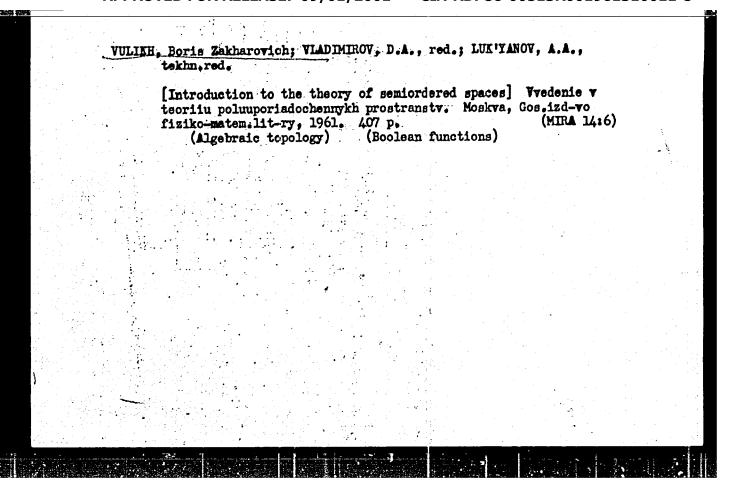
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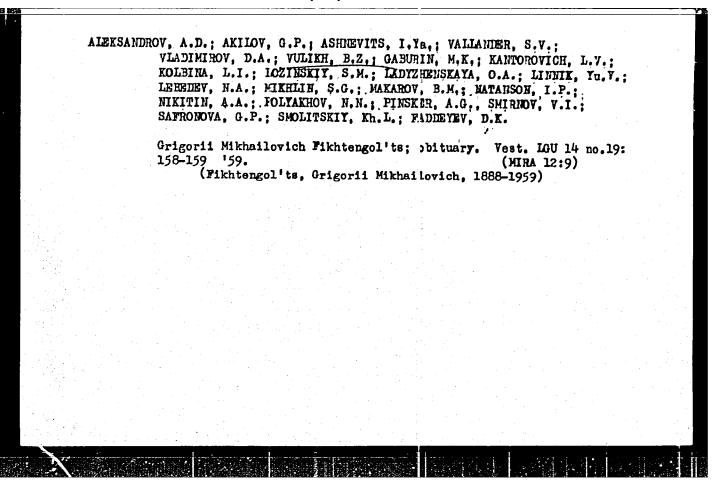
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